

## PATENT COOPERATION TREATY

## PCT

## INTERNATIONAL SEARCH REPORT

(PCT Article 18 and Rules 43 and 44)

Applicant's or agent's file reference <b>J742-PCT</b>	<b>FOR FURTHER ACTION</b> see Notification of Transmittal of International Search Report (Form PCT/ISA/220) as well as, where applicable, item 5 below.	
International application No. <b>PCT/JP 01/ 03537</b>	International filing date (day/month/year) <b>24/04/2001</b>	(Earliest) Priority Date (day/month/year) <b>28/04/2000</b>
Applicant <b>SHOWA DENKO K. K. et al.</b>		

This International Search Report has been prepared by this International Searching Authority and is transmitted to the applicant according to Article 18. A copy is being transmitted to the International Bureau.

This International Search Report consists of a total of 3 sheets.

It is also accompanied by a copy of each prior art document cited in this report.

**1. Basis of the report**

a. With regard to the **language**, the international search was carried out on the basis of the international application in the language in which it was filed, unless otherwise indicated under this item.

the international search was carried out on the basis of a translation of the international application furnished to this Authority (Rule 23.1(b)).

b. With regard to any **nucleotide and/or amino acid sequence** disclosed in the international application, the international search was carried out on the basis of the sequence listing :

contained in the international application in written form.

filed together with the international application in computer readable form.

furnished subsequently to this Authority in written form.

furnished subsequently to this Authority in computer readable form.

the statement that the subsequently furnished written sequence listing does not go beyond the disclosure in the international application as filed has been furnished.

the statement that the information recorded in computer readable form is identical to the written sequence listing has been furnished

2.  **Certain claims were found unsearchable** (See Box I).

3.  **Unity of invention is lacking** (see Box II).

4. With regard to the **title**,

the text is approved as submitted by the applicant.

the text has been established by this Authority to read as follows:

5. With regard to the **abstract**,

the text is approved as submitted by the applicant.

the text has been established, according to Rule 38.2(b), by this Authority as it appears in Box III. The applicant may, within one month from the date of mailing of this international search report, submit comments to this Authority.

6. The figure of the drawings to be published with the abstract is Figure No. \_\_\_\_\_

as suggested by the applicant.

because the applicant failed to suggest a figure.

because this figure better characterizes the invention.

None of the figures.

## INTERNATIONAL SEARCH REPORT

International Application No

PCT/JP 01/03537

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C08G65/18 C08G59/68

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)  
IPC 7 C08G

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, PAJ, WPI Data, CHEM ABS Data

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 3 457 193 A (TINSLEY SAMUEL W ET AL) 22 July 1969 (1969-07-22)  example 18 ---	1,2,4,5, 12,13, 15,16, 23,26
A	GB 2 310 211 A (KANSAI PAINT CO LTD) 20 August 1997 (1997-08-20) claims 1,5; table 1 ---	1-28
A	PATENT ABSTRACTS OF JAPAN vol. 1999, no. 08, 30 June 1999 (1999-06-30) & JP 11 061034 A (KANSAI PAINT CO LTD), 5 March 1999 (1999-03-05) abstract --- -/-	1-28

 Further documents are listed in the continuation of box C. Patent family members are listed in annex.

## \* Special categories of cited documents:

- \*A\* document defining the general state of the art which is not considered to be of particular relevance
- \*E\* earlier document but published on or after the international filing date
- \*L\* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- \*O\* document referring to an oral disclosure, use, exhibition or other means
- \*P\* document published prior to the international filing date but later than the priority date claimed

- \*T\* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- \*X\* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- \*Y\* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- \*&\* document member of the same patent family

Date of the actual completion of the international search	Date of mailing of the international search report
13 December 2001	02/01/2002

Name and mailing address of the ISA  
 European Patent Office, P.B. 5818 Patentlaan 2  
 NL - 2280 HV Rijswijk  
 Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,  
 Fax: (+31-70) 340-3016

Authorized officer

O'Sullivan, T

## INTERNATIONAL SEARCH REPORT

International Application No

PCT/JP 01/03537

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 897 710 A (ESPE DENTAL AG) 24 February 1999 (1999-02-24) claim 1 page 4 ---	1-28
A	US 3 388 105 A (JOACHIM GALINKE ET AL) 11 June 1968 (1968-06-11) cited in the application claim 1 ---	1-28
A	DE 196 47 848 A (TOAGOSEI CO) 22 May 1997 (1997-05-22) claim 1 ---	1-28
A	US 5 674 922 A (IGARASHI ICHIRO ET AL) 7 October 1997 (1997-10-07) claims 1-5 column 6, line 40 - line 50 -----	1-28

## INTERNATIONAL SEARCH REPORT

International Application No

PCT/JP 01/03537

Patent document cited in search report		Publication date		Patent family member(s)		Publication date
US 3457193	A	22-07-1969	NONE			
GB 2310211	A	20-08-1997	JP US	9278866 A 5882842 A	28-10-1997 16-03-1999	
JP 11061034	A	05-03-1999	CA EP US US	2223392 A1 0846739 A2 6262147 B1 6015848 A	05-06-1998 10-06-1998 17-07-2001 18-01-2000	
EP 0897710	A	24-02-1999	DE AU CA CN EP JP US	19736471 A1 8083898 A 2245548 A1 1209313 A 0897710 A2 11130945 A 6084004 A	25-02-1999 04-03-1999 21-02-1999 03-03-1999 24-02-1999 18-05-1999 04-07-2000	
US 3388105	A	11-06-1968	DE FR	1301132 B 1463054 A		03-03-1967
DE 19647848	A	22-05-1997	JP JP DE	3047800 B2 9143259 A 19647848 A1	05-06-2000 03-06-1997 22-05-1997	
US 5674922	A	07-10-1997	JP JP	3161583 B2 9031186 A	25-04-2001 04-02-1997	

## PATENT COOPERATION TREATY

From the INTERNATIONAL SEARCHING AUTHORITY

PCT

To:  
 A. AOKI, ISHIDA, & ASSOCIATES  
 Attn. Ishida, Takashi  
 Toranomon 37 Mori Bldg., 5-1,  
 Toranomon 3-chome, Minato-Ku  
 TOKYO 105-8423  
 JAPAN

NOTIFICATION OF TRANSMITTAL OF  
THE INTERNATIONAL SEARCH REPORT  
OR THE DECLARATION

(PCT Rule 44.1)

Date of mailing  
(day/month/year)

02/01/2002

Applicant's or agent's file reference

J742-PCT

## FOR FURTHER ACTION

See paragraphs 1 and 4 below

International application No.

PCT/JP 01/03537

International filing date

(day/month/year)

24/04/2001

Applicant

SHOWA DENKO K. K. et al.

1.  The applicant is hereby notified that the International Search Report has been established and is transmitted herewith.

**Filing of amendments and statement under Article 19:**

The applicant is entitled, if he so wishes, to amend the claims of the International Application (see Rule 46):

**When?** The time limit for filing such amendments is normally 2 months from the date of transmittal of the International Search Report; however, for more details, see the notes on the accompanying sheet.

**Where?** Directly to the International Bureau of WIPO  
 34, chemin des Colombettes  
 1211 Geneva 20, Switzerland  
 Fascimile No.: (41-22) 740.14.35

For more detailed instructions, see the notes on the accompanying sheet.

2.  The applicant is hereby notified that no International Search Report will be established and that the declaration under Article 17(2)(a) to that effect is transmitted herewith.

3.  **With regard to the protest** against payment of (an) additional fee(s) under Rule 40.2, the applicant is notified that:

the protest together with the decision thereon has been transmitted to the International Bureau together with the applicant's request to forward the texts of both the protest and the decision thereon to the designated Offices.

no decision has been made yet on the protest; the applicant will be notified as soon as a decision is made.

4. **Further action(s):** The applicant is reminded of the following:

Shortly after **18 months** from the priority date, the international application will be published by the International Bureau. If the applicant wishes to avoid or postpone publication, a notice of withdrawal of the international application, or of the priority claim, must reach the International Bureau as provided in Rules 90bis.1 and 90bis.3, respectively, before the completion of the technical preparations for international publication.

Within **19 months** from the priority date, a demand for international preliminary examination must be filed if the applicant wishes to postpone the entry into the national phase until 30 months from the priority date (in some Offices even later).

Within **20 months** from the priority date, the applicant must perform the prescribed acts for entry into the national phase before all designated Offices which have not been elected in the demand or in a later election within 19 months from the priority date or could not be elected because they are not bound by Chapter II.

Name and mailing address of the International Searching Authority  
 European Patent Office, P.B. 5818 Patentlaan 2  
 NL-2280 HV Rijswijk  
 Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,  
 Fax: (+31-70) 340-3016

Authorized officer

Alfredo Prein

These Notes are intended to give the basic instructions concerning the filing of amendments under article 19. The Notes are based on the requirements of the Patent Cooperation Treaty, the Regulations and the Administrative Instructions under that Treaty. In case of discrepancy between these Notes and those requirements, the latter are applicable. For more detailed information, see also the PCT Applicant's Guide, a publication of WIPO.

In these Notes, "Article", "Rule", and "Section" refer to the provisions of the PCT, the PCT Regulations and the PCT Administrative Instructions respectively.

## INSTRUCTIONS CONCERNING AMENDMENTS UNDER ARTICLE 19

The applicant has, after having received the international search report, one opportunity to amend the claims of the international application. It should however be emphasized that, since all parts of the international application (claims, description and drawings) may be amended during the international preliminary examination procedure, there is usually no need to file amendments of the claims under Article 19 except where, e.g. the applicant wants the latter to be published for the purposes of provisional protection or has another reason for amending the claims before international publication. Furthermore, it should be emphasized that provisional protection is available in some States only.

### What parts of the International application may be amended?

Under Article 19, only the claims may be amended.

During the international phase, the claims may also be amended (or further amended) under Article 34 before the International Preliminary Examining Authority. The description and drawings may only be amended under Article 34 before the International Examining Authority.

Upon entry into the national phase, all parts of the international application may be amended under Article 28 or, where applicable, Article 41.

**When?** Within 2 months from the date of transmittal of the international search report or 16 months from the priority date, whichever time limit expires later. It should be noted, however, that the amendments will be considered as having been received on time if they are received by the International Bureau after the expiration of the applicable time limit but before the completion of the technical preparations for international publication (Rule 46.1).

### Where not to file the amendments?

The amendments may only be filed with the International Bureau and not with the receiving Office or the International Searching Authority (Rule 46.2).

Where a demand for international preliminary examination has been/is filed, see below.

**How?** Either by cancelling one or more entire claims, by adding one or more new claims or by amending the text of one or more of the claims as filed.

A replacement sheet must be submitted for each sheet of the claims which, on account of an amendment or amendments, differs from the sheet originally filed.

All the claims appearing on a replacement sheet must be numbered in Arabic numerals. Where a claim is cancelled, no renumbering of the other claims is required. In all cases where claims are renumbered, they must be renumbered consecutively (Administrative Instructions, Section 205(b)).

**The amendments must be made in the language in which the International application is to be published.**

### What documents must/may accompany the amendments?

**Letter (Section 205(b)):**

The amendments must be submitted with a letter.

The letter will not be published with the international application and the amended claims. It should not be confused with the "Statement under Article 19(1)" (see below, under "Statement under Article 19(1)").

**The letter must be in English or French, at the choice of the applicant. However, if the language of the International application is English, the letter must be in English; if the language of the International application is French, the letter must be in French.**

## NOTES TO FORM PCT/ISA/220 (cont'd)

The letter must indicate the differences between the claims as filed and the claims as amended. It must, in particular, indicate, in connection with each claim appearing in the international application (it being understood that identical indications concerning several claims may be grouped), whether

- (i) the claim is unchanged;
- (ii) the claim is cancelled;
- (iii) the claim is new;
- (iv) the claim replaces one or more claims as filed;
- (v) the claim is the result of the division of a claim as filed.

**The following examples illustrate the manner in which amendments must be explained in the accompanying letter:**

1. [Where originally there were 48 claims and after amendment of some claims there are 51]:  
"Claims 1 to 29, 31, 32, 34, 35, 37 to 48 replaced by amended claims bearing the same numbers; claims 30, 33 and 36 unchanged; new claims 49 to 51 added."
2. [Where originally there were 15 claims and after amendment of all claims there are 11]:  
"Claims 1 to 15 replaced by amended claims 1 to 11."
3. [Where originally there were 14 claims and the amendments consist in cancelling some claims and in adding new claims]:  
"Claims 1 to 6 and 14 unchanged; claims 7 to 13 cancelled; new claims 15, 16 and 17 added." or  
"Claims 7 to 13 cancelled; new claims 15, 16 and 17 added; all other claims unchanged."
4. [Where various kinds of amendments are made]:  
"Claims 1-10 unchanged; claims 11 to 13, 18 and 19 cancelled; claims 14, 15 and 16 replaced by amended claim 14; claim 17 subdivided into amended claims 15, 16 and 17; new claims 20 and 21 added."

### **"Statement under article 19(1)" (Rule 46.4)**

The amendments may be accompanied by a statement explaining the amendments and indicating any impact that such amendments might have on the description and the drawings (which cannot be amended under Article 19(1)).

The statement will be published with the international application and the amended claims.

**It must be in the language in which the international application is to be published.**

It must be brief, not exceeding 500 words if in English or if translated into English.

It should not be confused with and does not replace the letter indicating the differences between the claims as filed and as amended. It must be filed on a separate sheet and must be identified as such by a heading, preferably by using the words "Statement under Article 19(1)."

It may not contain any disparaging comments on the international search report or the relevance of citations contained in that report. Reference to citations, relevant to a given claim, contained in the international search report may be made only in connection with an amendment of that claim.

### **Consequence if a demand for international preliminary examination has already been filed**

If, at the time of filing any amendments under Article 19, a demand for international preliminary examination has already been submitted, the applicant must preferably, at the same time of filing the amendments with the International Bureau, also file a copy of such amendments with the International Preliminary Examining Authority (see Rule 62.2(a), first sentence).

### **Consequence with regard to translation of the international application for entry into the national phase**

The applicant's attention is drawn to the fact that, where upon entry into the national phase, a translation of the claims as amended under Article 19 may have to be furnished to the designated/elected Offices, instead of, or in addition to, the translation of the claims as filed.

For further details on the requirements of each designated/elected Office, see Volume II of the PCT Applicant's Guide.

(19) World Intellectual Property Organization  
International Bureau(43) International Publication Date  
8 November 2001 (08.11.2001)

PCT

(10) International Publication Number  
WO 01/83580 A2(51) International Patent Classification<sup>7</sup>: C08G 59/00 (74) Agents: ISHIDA, Takashi et al.; A. AOKI, ISHIDA & ASSOCIATES, Toranomon 37 Mori Bldg., 5-1, Toranomon 3-chome, Minato-ku, Tokyo 105-8423 (JP).

(21) International Application Number: PCT/JP01/03537

(22) International Filing Date: 24 April 2001 (24.04.2001)

(81) Designated States (national): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, KE, KG, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW.

(25) Filing Language: English

(26) Publication Language: English

(84) Designated States (regional): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).

(30) Priority Data:  
2000-130472 28 April 2000 (28.04.2000) JP  
60/263,163 23 January 2001 (23.01.2001) US

(71) Applicant (for all designated States except US): SHOWA DENKO K. K. [JP/JP]; 13-9, Shiba Daimon 1-chome, Minato-ku, Tokyo 105-8518 (JP).

## Published:

— without international search report and to be republished upon receipt of that report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(72) Inventors; and  
(75) Inventors/Applicants (for US only): WATANABE, Takeo [JP/JP]; C/O CENTRAL RESEARCH LABORATORY, SHOWA DENKO K.K., 1-1, Ohnodai 1-chome, Midori-ku, Chiba-shi, Chiba 267-0056 (JP). SATO, Takashi [JP/JP]; C/O CENTRAL RESEARCH LABORATORY, SHOWA DENKO K.K., 1-1, Ohnodai 1-chome, Midori-ku, Chiba-shi, Chiba 267-0056 (JP). TAGOSHI, Hirotaka [JP/JP]; C/O CENTRAL RESEARCH LABORATORY, SHOWA DENKO K.K., 1-1, Ohnodai 1-chome, Midori-ku, Chiba-shi, Chiba 267-0056 (JP).

WO 01/83580 A2

(54) Title: POLYMERIZABLE COMPOSITION, CURED MATERIAL THEREOF AND METHOD FOR MANUFACTURING THE SAME

(57) Abstract: The present invention provides a polymerizable composition comprising (a) an alicyclic alkane having at least one oxetanyl group and at least one epoxy group within the same molecule and (b) a compound capable of initiating cationic polymerization under irradiation of an active energy ray and/or under heat exhibits high activity in the photo- or heat-cationic ring-opening polymerization exhibiting high activity, e.g., rapid polymerizability, rapid curability, under irradiation of an active energy ray and/or under heat. The present invention also provides a polymerization product thereof.

## DESCRIPTION

Polymerizable Composition, Cured Material Thereof  
and Method for Manufacturing the Same

5

## Cross Reference to Related Application

This application is based on U.S. Provisional Application No. 60/263,163, filed on January 23, 2001.

## 10 Technical Field

The present invention relates to a polymerizable composition which rapidly polymerizes under irradiation of an active energy ray such as ultraviolet light or an electron beam and/or under heat to provide a cured material, a method for polymerizing the composition and a cured material obtained by polymerizing the composition.

## Background Art

Curing of a resin by an active energy ray such as ultraviolet light is favored with various characteristics such as, for example, a high curing rate, good workability due to the non-use of a solvent, and a requirement for an extremely low energy amount, and therefore this technique is playing a very important role in various fields such as wood coating, metal painting and printing. At the initial stage of development in these fields, studies were exclusively directed to ultraviolet light initiation radical polymerization of polyfunctional acrylates, unsaturated polyesters or the like, and various materials comprising these polyfunctional acrylates, unsaturated polyesters or the like have been heretofore used. Even at the present date, most studies are directed to ultraviolet light initiation radical polymerization. However, it is now recognized that photoinitiation ionic polymerization is also fairly promising in various applied fields. The photoinitiation ionic polymerization, in which various

monomers can be used, has the possibility of providing cured materials having various chemical and physical properties, and therefore this technique is attracting attention.

5

The photocationic polymerizable composition comprises a photocationic polymerization initiator (photoacid generator) capable of decomposing, and thereby generating an acid under irradiation of an active energy ray such as ultraviolet light, and a compound capable of causing a polymerization reaction or a crosslinking reaction by the generated acid. The development thereof is exclusively directed to epoxy resins having an oxirane ring which is a 3-membered cyclic ether, and the photocurable epoxy resins are known to have excellent properties in adhesion, heat resistance and chemicals resistance. However, conventional photocurable epoxy resins have a problem in that their photocuring rates are very low. Therefore, improving the curing rate while maintaining the properties of the epoxy resin is required.

A polyfunctional oxetane monomer comprising, as its polymerizable group, more than one oxetane ring, i.e. four-membered cyclic ethers, per molecule is reported to have photocurability equal to or higher than that of the corresponding polyfunctional epoxides (see, Journal of Macromolecular Science, Vol. A 29, No. 10, page 915 (1992); ibid., Vol. 30, Nos. 2 & 3, page 173; ibid., Vol. 30, Nos. 2 & 3, page 189 (1993)). Also, a photocurable composition comprising a polyfunctional oxetane monomer as a main component had been proposed (see, JP-A (Japanese Unexamined Patent Publication (Kokai)) No. 6-16804). The ultraviolet light curable resin using this polyfunctional oxetane monomer shows a high curing rate under irradiation by ultraviolet light as compared with epoxy resins, however, it encounters difficulties in

application to uses such as coating on a material having low heat resistance, e.g. paper or plastic. Also, a higher photocuring rate is required.

5 . The photocationic polymerization curable epoxy resin composition or the photocationic polymerization curable oxetane compound-containing resin composition may also be thermopolymerized by adding a heat cationic polymerization initiator which decomposes under heat to 10 generate an acid, in place of adding a photocationic polymerization initiator. The photocationic polymerization initiator accelerates the ring-opening cationic polymerization of a cationic polymerizable compound under irradiation of an active energy ray. On 15 the contrary, the heat cationic polymerization initiator accelerates the ring-opening cationic polymerization when the compound is externally heated or left at a room temperature, whereby this technique is particularly important in the field where the photocationic 20 polymerization cannot be performed, for example, in curing a material having poor light transmittance or in manufacturing a thick cast molded article. Regarding heat cationic polymerization, see JP-A NO.11-246541, JP-A NO.11-106380 and JP-A NO.11-61034, etc. However, 25 similarly to the photocationic polymerization, the curing rate of the heat cationic polymerization curable epoxy resin composition, the heat cationic polymerization curable oxetane compound-containing resin composition or mixture thereof is not sufficiently high, and therefore 30 in order to obtain a cured material, a high temperature treatment for a long period of time or a large amount of heat cationic polymerization initiator is required. Accordingly, a composition satisfying productivity and profitability has not yet been obtained.

35

U.S. Patent No.3,388,105 describes a technique of curing an alicyclic alkane, having an oxetanyl group and

an epoxy group within the same molecule, by addition-reacting it under heat with a carboxyl group-containing compound. However, it is not known whether this alicyclic alkane can exhibit very high activity  
5 (curability) in the cationic ring-opening polymerization.

#### Disclosure of the Invention

Under these circumstances, the present invention has been accomplished, and the object of the present  
10 invention is to provide a cationic polymerizable composition capable of exhibiting high activity (rapid polymerizability, rapid curability) under irradiation of an active energy ray and/or under heat.

15 As a result of extensive investigations to solve the above-described problems, the present inventors have found that a polymerizable composition comprising (a) an alicyclic alkane having at least one oxetanyl group and at least one epoxy group within the same molecule and (b)  
20 a compound capable of initiating cationic polymerization under irradiation of an active energy ray and/or under heat exhibits high activity (rapid polymerizability, rapid curability) in the cationic ring-opening polymerization under irradiation of an active energy ray and/or under heat, and at the same time, found that the  
25 polymerization product thereof (cured material) has good properties. The present invention has been accomplished based on these findings.

30 More specifically, the present invention relates to a polymerizable composition, a cured material obtained by polymerizing the composition and a method for manufacturing the cured material, described in the followings (1) to (28).

35

(1) A polymerizable composition comprising (a) an alicyclic alkane having at least one oxetanyl group and

at least one epoxy group within the same molecule and (b) a compound capable of initiating cationic polymerization under irradiation of an active energy ray and/or under heat.

5

(2) The polymerizable composition according to (1), wherein said compound (b) is a compound capable of generating an acid under irradiation of an active energy ray and/or under heat, and thereby initiating the cationic polymerization.

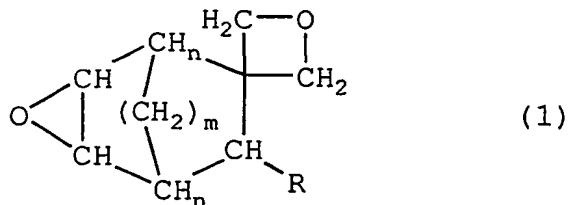
10

(3) The polymerizable composition according to (2), wherein said compound (b) is one or more compound selected from the group consisting of sulfonium salts, iodonium salts and diazonium salts.

15

(4) The polymerizable composition according to (1), wherein said alicyclic alkane (a) is a compound represented by formula (1):

20



wherein R represents a hydrogen atom or a methyl group, m represents an integer of 0 to 2, and n is 2 when m is 0 and otherwise n is 1.

25

(5) The polymerizable composition according to (4), wherein said alicyclic alkane (a) is 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane.

30

(6) The polymerizable composition according to (1), which comprises a compound (c) that can be cationic polymerized by said compound (b), wherein said compound (c) is other than said compound (a).

5 (7) The polymerizable composition according to (6), wherein at least one of said compound (c) is a compound (c-1) having one or more epoxy group that can be cationic polymerized by said compound (b).

10 (8) The polymerizable composition according to (6), wherein at least one of said compound (c) is a compound (c-2) having one or more oxetanyl group that can be cationic polymerized by said compound (b).

15 (9) The polymerizable composition according to (1), wherein said alicyclic alkane (a) is blended in an amount of 5 to 100 mass % based on the total mass of the polymerizable composition minus the mass of said compound (b).

20 (10) The polymerizable composition according to (7), wherein said compound (c-1) is blended in an amount of 5 to 95 mass % based on the total mass of the polymerizable composition minus the mass of said compound (b).

25 (11) The polymerizable composition according to (8), wherein said compound (c-2) is blended in an amount of 5 to 95 mass % based on the total mass of the polymerizable composition minus the mass of said compound (b).

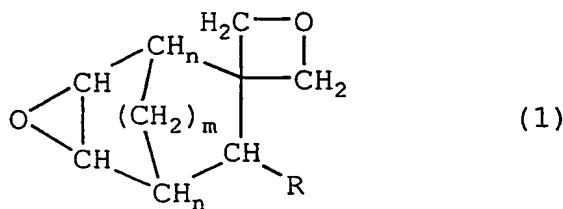
30 (12). A cured material obtained by polymerizing a polymerizable composition comprising (a) an alicyclic alkane having at least one oxetanyl group and at least one epoxy group within the molecule and (b) a compound capable of initiating cationic polymerization under irradiation of an active energy ray and/or under heat.

35 (13) The cured material according to (12), wherein said compound (b) is a compound capable of generating an acid under irradiation of an active energy ray and/or

under heat, and thereby initiating the cationic polymerization.

5 (14) The cured material according to (13), wherein said compound (b) is one or more compound selected from the group consisting of sulfonium salts, iodonium salts and diazonium salts.

10 (15) The cured material according to (12), wherein said alicyclic alkane (a) is a compound represented by formula (1):



15 wherein R represents a hydrogen atom or a methyl group, m represents an integer of 0 to 2, and n is 2 when m is 0 and otherwise n is 1.

20 (16) The cured material according to (15), wherein said alicyclic alkane (a) is 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane.

25 (17) The cured material according to (16), which comprises a compound (c) that can be cationic polymerized by said compound (b), wherein said compound (c) is other than said compound (a).

30 (18) The cured material according to (17), wherein at least one of said compound (c) is a compound (c-1) having one or more epoxy group that can be cationic polymerized by said compound (b).

(19) The cured material according to (17), wherein at least one of said compound (c) is a compound (c-2)

having one or more oxetanyl group that can be cationic polymerized by said compound (b).

5 (20) The cured material according to (12), wherein said alicyclic alkane (a) is blended in an amount of 5 to 100 mass% based on the total mass of the polymerizable composition minus the mass of said compound (b).

10 (21) The cured material according to (18), wherein said compound (c-1) is blended in an amount of 5 to 95 mass % based on the total mass of the polymerizable composition minus the mass of said compound (b).

15 (22) The cured material according to (19), wherein said compound (c-2) is blended in an amount of 5 to 95 mass % based on the total mass of the polymerizable composition minus the mass of said compound (b).

20 (23) A method for manufacturing a cured material from a polymerizable composition comprising (a) an alicyclic alkane having at least one oxetanyl group and at least one epoxy group within the same molecule and (b) a compound capable of initiating cationic polymerization under irradiation of an active energy ray and/or under 25 heat, said method comprising initiating the polymerization under irradiation of an active energy ray and/or under heat.

30 (24) A method for manufacturing the cured material according to any of (12) to (22), comprising initiating the polymerization under irradiation of an active energy ray.

35 (25) The method according to (24), wherein said active ray is ultraviolet light.

(26) A method for manufacturing the cured material

according to any of (12) to (22), comprising initiating the polymerization under heat.

5 (27) A method for manufacturing the cured material according to any of (12) to (22), comprising initiating the polymerization under irradiation of an active energy ray and then heating the composition.

10 (28) The method according to (27), wherein said active ray is ultraviolet light.

#### Best Mode for Carrying Out the Invention

The present invention is described in detail below.

15 The present invention relates to a polymerizable composition comprising (a) an alicyclic alkane having at least one oxetanyl group and at least one epoxy group within the same molecule and (b) a compound capable of initiating cationic polymerization under irradiation of 20 an active energy ray and/or under heat, and also relates to a method for polymerizing the composition and a cured material of the composition.

25 Examples of the alicyclic alkane (a) having at least one oxetanyl group and at least one epoxy group within the same molecule for use in the present invention include 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane, 6,7-epoxy-2-oxaspiro[3.5]nonane, 7,8-epoxy-2-oxa-5-methylspiro[3.6]decane, 5,6-epoxy-2-oxaspiro[3.6]decane, 30 spiro[5,6-epoxynorbornane-2,3'-oxetane] and spiro[5,6-epoxy-3-methylnorbornane-2,3'-oxetane]. Among these, preferred are 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane, 6,7-epoxy-2-oxaspiro[3.5]nonane and spiro[5,6-epoxynorbornane-2,3'-oxetane].

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Alicyclic alkanes having at least one oxetanyl group and at least one epoxy group within the same molecule can

be easily synthesized by a known method described, for example, in U.S. Patent 3,388,105.

5        Alicyclic alkanes (a) having at least one oxetanyl group and at least one epoxy group within the same molecule can be used individually or as a mixture of two or more thereof.

10       The compound (a) is suitably blended in an amount of from 5 to 100 mass % (in the case of using two or more compounds (a), the total amount thereof) based on the total weight of the polymerizable composition of the present invention minus the weight of the compound (b) capable of initiating cationic polymerization under 15       irradiation of an active energy ray and/or under heat.

20       The compound (b) capable of initiating cationic polymerization under irradiation of an active energy ray and/or under heat for use in the present invention may be a compound which is subjected to a conversion under heat or under irradiation of an active energy ray such as ultraviolet light, and thereby generates a substance capable of initiating the cationic polymerization, such as an acid. Accordingly, the compound (b) is one of a 25       cationic polymerization initiator and is called an "acid generator" in the art. The compound (b) is hereinafter referred to as an acid generation-type cationic polymerization initiator.

30       The acid generation-type cationic polymerization initiator is added in order to accelerate the ring-opening cationic polymerization of both the oxetanyl group and the epoxy group in the alicyclic alkane (a) having at least one oxetanyl group and at least one epoxy group within the same molecule of the present invention, under heat or irradiation with light such as ultraviolet light, and to allow smooth curing of the curable material 35

or coating to proceed.

The acid generation-type cationic polymerization initiator as used in the present invention is a compound which is subjected to a conversion under heat or under irradiation of an active energy ray such as ultraviolet light and thereby generates a substance capable of initiating the cationic polymerization, such as an acid, and excludes compounds originally in an acid form, such as carboxylic acid.

Examples of the acid generation-type cationic polymerization initiator include known sulfonium salts, iodonium salts, phosphonium salts, diazonium salts, ammonium salts, ferrocenes and the like. Specific examples thereof are described below, however, the present invention is not limited to these compounds.

Examples of the sulfonium salt-based acid generation-type cationic polymerization initiator include bis[4-(diphenylsulfonio)phenyl]sulfide bishexafluorophosphate, bis[4-(diphenylsulfonio)phenyl]sulfide bishexafluoro-antimonate, bis[4-(diphenylsulfonio)phenyl]sulfide bistetrafluoroborate, bis[4-(diphenylsulfonio)phenyl]-sulfide tetrakis(pentafluorophenyl)borate, diphenyl-4-(phenylthio)phenylsulfonium hexafluorophosphate, diphenyl-4-(phenylthio)phenylsulfonium hexafluoroantimonate, diphenyl-4-(phenylthio)phenylsulfonium tetrafluoroborate, diphenyl-4-(phenylthio)phenylsulfonium tetrakis(penta-fluorophenyl)borate, triphenylsulfonium hexafluorophosphate, triphenylsulfonium hexafluoroantimonate, triphenylsulfonium tetrafluoroborate, triphenylsulfonium tetrakis(pentafluorophenyl)borate, bis[4-(di(4-(2-hydroxyethoxy)phenylsulfonio)phenyl]sulfide

5 bishexafluorophosphate, bis[4-(di(4-(2-hydroxyethoxy)phenylsulfonio)phenyl)sulfide bis-hexafluoroantimonate, bis[4-(di(4-(2-hydroxyethoxy)phenylsulfonio)phenyl)sulfide bistetrafluoroborate and  
bis[4-(di(4-(2-hydroxyethoxy)phenylsulfonio)phenyl)sulfide tetrakis(pentafluorophenyl)borate.

10 Examples of the iodonium salt-based acid generation-type cationic polymerization initiator include diphenyl-iodonium hexafluorophosphate, diphenyliodonium hexafluoroantimonate, diphenyliodonium tetrafluoroborate, diphenyliodonium tetrakis(pentafluorophenyl)borate, bis(dodecylphenyl)iodonium hexafluorophosphate, bis(dodecyl-15 phenyl)iodonium hexafluoroantimonate, bis(dodecylphenyl)-iodonium tetrafluoroborate, bis(dodecylphenyl)iodonium tetrakis(pentafluorophenyl)borate, 4-methylphenyl-4-(1-methylethyl)phenyliodonium hexafluorophosphate, 4-methylphenyl-4-(1-methylethyl)phenyliodonium hexafluoro-20 antimonate, 4-methylphenyl-4-(1-methylethyl)phenyliodonium tetrafluoroborate and 4-methylphenyl-4-(1-methylethyl)phenyliodonium tetrakis(pentafluorophenyl)borate.

25 Examples of the phosphonium salt-based acid generation-type cationic polymerization initiator include ethyltriphenyl phosphonium tetrafluoroborate, ethyltriphenyl phosphonium hexafluorophosphate, ethyltriphenyl phosphonium hexafluoroantimonate, 30 tetrabutyl phosphonium tetrafluoroborate, tetrabutyl phosphonium hexafluorophosphate, tetrabutyl phosphonium hexafluoroantimonate.

35 Examples of the diazonium salt-based acid generation-type cationic polymerization initiator include phenyldiazonium hexafluorophosphate, phenyldiazonium hexafluoroantimonate, phenyldiazonium tetrafluoroborate

and phenyldiazonium tetrakis(pentafluorophenyl)borate.

5 Examples of the ammonium salt-type acid generation-type cationic polymerization initiator include 1-benzyl-2-cyanopyridinium hexafluorophosphate, 1-benzyl-2-cyano-pyridinium hexafluoroantimonate, 1-benzyl-2-cyanopyridinium tetrafluoroborate, 1-benzyl-2-cyanopyridinium tetrakis(pentafluorophenyl)borate, 1-(naphthylmethyl)-2-cyanopyridinium hexafluorophosphate, 10 1-(naphthylmethyl)-2-cyanopyridinium hexafluoroantimonate, 1-(naphthylmethyl)-2-cyanopyridinium tetrafluoroborate and 1-(naphthylmethyl)-2-cyanopyridinium tetrakis(pentafluorophenyl)borate.

15 Examples of the ferrocene-based acid generation-type cationic polymerization initiator include, but not limited to, (2,4-cyclo-pentadien-1-yl)[(1-methylethyl)benzene]-Fe(II) hexafluorophosphate, (2,4-cyclopentadien-1-yl)[(1-methylethyl)benzene]-Fe(II) hexafluoroantimonate, (2,4-cyclopentadien-1-yl)[(1-methylethyl)benzene]-Fe(II) tetrafluoroborate and (2,4-cyclopentadien-1-yl)[(1-methylethyl)benzene]-Fe(II) tetrakis(pentafluorophenyl)borate.

25 Among these, sulfonium salt-based and iodonium salt-based acid generation-type cationic polymerization initiators are preferred in view of curing rate, stability and profitability. Examples of the commercially available product include SP-150, SP-170, 30 CP-66 and CP-77 produced by Asahi Denka Kogyo; CYRACURE-UVI-6990 and UVI-6974 produced by Union Carbide; CI-2855 and CI-2639 produced by Nippon Soda; San-Aid SI-60 produced by Sanshin Kagaku Kogyo K.K.; IRGACURE 261 ((2,4-cyclopentadien-1-yl) [(1-methylethyl)benzene]-Fe(II) hexafluorophosphate) produced by Ciba Specialty 35 Chemicals; and RHODORSIL 2074 (4-methylphenyl-4-(1-methylethyl)phenyliodonium tetrakis-

(pentafluorophenyl)borate) produced by Rhone Poulen.

The acid generation-type cationic polymerization initiator may be selected from the above-described materials and these materials may be used individually or in a combination of two or more thereof. The suitable range of the amount of the acid generation-type cationic polymerization initiator is not critical, but is preferably used in an amount of 0.05 to 25 parts by mass, preferably from 1 to 10 parts by mass, based on the total amount (100 parts by mass) of the polymerizable compounds (namely, the alicyclic alkane (a) having at least one oxetanyl group and at least one epoxy group within the same molecule and the compound (c)) which can be cationic polymerized by the compound (b) capable of initiating cationic polymerization under irradiation of an active energy ray and/or under heat. If adding it in an amount less than 0.05 parts by mass, poor sensitivity results, and a tremendously large amount of light irradiation energy or a high-temperature treatment over an extended time period will be required for curing. On the other hand, even though adding it in an amount more than 25 parts by mass, the sensitivity will not increase and this is not preferred also in view of the profitability. Moreover, the amount of the initiator remaining in the coating as an uncured component will increase, and thereby the physical properties of the cured product may deteriorate.

For the compound (c-1) having an epoxy group for use in the present invention, an epoxy compound commonly known and used can be employed, and this compound is not particularly limited as long as it has one or more epoxy group within one molecule.

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Specific examples of the compound which can be used include bisphenol A diglycidyl ether, bisphenol F

5        diglycidyl ether, bisphenol S diglycidyl ether, brominated bisphenol A diglycidyl ether, brominated bisphenol F diglycidyl ether, brominated bisphenol S diglycidyl ether, epoxy novolak resin, hydrogenated bisphenol A diglycidyl ether, hydrogenated bisphenol F diglycidyl ether, hydrogenated bisphenol S diglycidyl ether, cresol novolak epoxy resin and triglycidyl isocyanurate.

10       Examples of the alicyclic epoxy compound include (3,4-epoxycyclohexyl)methyl-3',4'-epoxycyclohexyl carboxylate, 2-(3,4-epoxycyclohexyl-5,5-spiro-3,4-epoxy)cyclohexane-meta-dioxane, bis(3,4-epoxycyclohexylmethyl) adipate, vinylcyclohexene oxide, 15       4-vinylepoxyhexane, bis(3,4-epoxy-6-methylcyclohexylmethyl) adipate, 3,4-epoxy-6-methylcyclohexyl-3',4'-epoxy-6'-methylcyclohexane carboxylate, methylenebis(3,4-epoxycyclohexane), dicyclopentadiene diepoxide, di(3,4-epoxycyclohexylmethyl) ether 20       of ethylene glycol, and methylenebis(3,4-epoxycyclohexane carboxylate).

25       Other examples include dioctyl epoxyhexahydrophthalate, di-2-ethylhexyl epoxyhexahydrophthalate, 1,4-butanedioldiglycidyl ether, 1,6-hexanedioldiglycidyl ether, glycerol triglycidyl ether, trimethylolpropane triglycidyl ether, polyethylene glycol diglycidyl ether, polypropylene glycol diglycidyl ether; polyglycidyl ethers of a polyether polyol obtained by 30       adding one or more alkylene oxide to an aliphatic polyhydric alcohol, such as ethylene glycol, propylene glycol and glycerol; diglycidyl ethers of aliphatic long-chained dibasic acid; monoglycidyl ethers of aliphatic higher alcohol; butyl glycidyl ether, phenyl glycidyl ether, cresol glycidyl ether, nonylphenyl glycidyl ether, 35       glycidyl methacrylate; monoglycidyl ethers of phenol, cresol, butylphenol or polyether alcohol obtained by

adding an alkylene oxide thereto; glycidyl esters of higher fatty acid; epoxidated soybean oil; butyl epoxystearate, octyl epoxystearate, epoxidated linseed oil and epoxidated polybutadiene.

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These compounds (c-1) having one or more epoxy group within the molecule can be used individually or as a mixture of two or more thereof.

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The compound (c-1) can be blended in an amount of 5 to 95 mass % (in the case of using two or more compounds (c-1), the total amount thereof) based on the total weight of the polymerizable composition of the present invention minus the weight of the compound (b) capable of initiating cationic polymerization under irradiation of an active energy ray and/or under heat.

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The compound (c-2) having one or more oxetanyl group within the molecule is described below. The compound (c-2) having one or more oxetanyl group is used for the purpose of adjusting the viscosity of the entire polymerizable composition according to its use end and also imparting effects considered to be ascribable to the oxetane compound, namely, water resistance or a reduction 25 in curing shrinkage.

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Specific examples thereof include trimethylene oxide, 3,3-dimethyloxetane, 3,3-dichloromethyloxetane, 3-ethyl-3-phenoxymethyloxetane, 3-ethyl-3-hydroxymethyloxetane (EOXA, trade name, produced by Toa Gosei), bis[(3-ethyl-3-oxetanylmethoxy)methyl]benzene (also called xylylene dioxetane; XDO, trade name, produced by Toa Gosei), tri[(3-ethyl-3-oxetanylmethoxy)methyl]benzene, bis[(3-ethyl-3-oxetanylmethoxy)methylphenyl]ether, (3-ethyl-3-oxetanylmethoxy)oligodimethylsiloxane, and compounds containing a multifunctional oxetane ring having a high

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molecular weight, specifically, oxetane oligomer (Oligo-OXT, trade name, produced by Toa Gosei)

Examples of the compound having an alicyclic structure other than an oxetane ring within the same compound include 2-oxaspiro[3.5]nonane, 7-methyl-2-oxaspiro[3.5]nonane, spiro[adamantane-2,3'-oxetane], spiro[bicyclo[2.2.1]heptane-2,3'-oxetane], spiro[bicyclo[2.2.2]octane-2,3'-oxetane], spiro[7-oxabicyclo[2.2.1]heptane-2,3'-oxetane], 2-oxaspiro[3.5]non-6-ene, 5-methyl-2-oxaspiro[3.5]non-6-ene, spiro[bicyclo[2.2.1]hept-5-ene-2,3'-oxetane], spiro[3-methylbicyclo[2.2.1]hept-5-ene-2,3'-oxetane], 5-methyl-2-oxaspiro[3.5]nonane and spiro[3-methylbicyclo[2.2.1]heptane-2,3'-oxetane].

These compounds (c-2) having one or more oxetanyl group can be used individually or as a mixture of two or more thereof.

The compound (c-2) can be blended in an amount of 5 to 95 mass % (in the case of using two or more compounds (c-2) in combination, the total amount thereof) based on the total weight of the polymerizable composition of the present invention minus the weight of the compound (b) capable of initiating cationic polymerization under irradiation of an active energy ray and/or under heat.

In the present invention, a cationic polymerizable monomer described below can also be added to the polymerizable composition. This cationic polymerizable monomer is classified into the compound (c) that causes a polymerization reaction or crosslinking reaction by the acid generated from the acid generation-type cationic polymerization initiator, but other than the compound (c-1) or (c-2). Examples thereof include oxolane compounds such as tetrahydrofuran, 2,3-dimethyl-tetrahydrofuran;

5       cyclic acetal compounds such as trioxan, 1,3-dioxolane and 1,3,6-trioxancyclooctane; cyclic lactone compounds such as  $\beta$ -propiolactone and  $\epsilon$ -caprolactone; thiirane compounds such as ethylenesulfide, 1,2-propylenesulfide and thioepichlorohydrin; thiethane compounds such as 3,3-dimethylthiethane; vinyl ether compounds such as ethylene glycol divinyl ether, triethylene glycol divinyl ether, trimethylol propane trivinyl ether; spiroorthoester compounds, a reaction product of an epoxy compound with 10 lactone; ethylenically unsaturated compounds such as vinylcyclohexane, isobutylene and polybutadiene; cyclic ether compounds; cyclic thioether compounds; and vinyl compounds.

15       These cationic polymerizable monomers can be added individually or in a combination of two or more thereof.

20       In polymerizing the polymerizable composition of the present invention by an ultraviolet light which is one kind of active energy rays, a sensitizer may also be used for improving the polymerization rate. Examples of the sensitizer used to this purpose include pyrene, perylene, 2,4-diethylthioxanthone, 2,4-dimethylthioxanthone, 2,4-dichlorothioxanthone and phenothiazine. In the case of 25 using a sensitizer, the sensitizer is preferably used in an amount of 0.1 to 100 parts by mass per 100 parts by mass of photoacid generation-type cationic polymerization initiator.

30       The polymerizable composition of the present invention may also contain an organic solvent. The organic solvent is used for the purpose of adjusting the viscosity according to the use or the coating method or in the case where the polymerizable composition is solid, 35 for dissolving and diluting the polymerizable composition and thereby enabling the coating of the polymerizable composition in a liquid form. Specific examples of the

organic solvent which can be used include known solvents such as acetone, hexane, ethyl acetate, diethyl ether, ethyl methyl ketone, cyclohexane, tetrahydrofuran, toluene, xylene, tetramethylbenzene, petroleum ether, 5 petroleum naphtha and solvent naphtha; ethylene glycol monoalkyl ethers and acetates thereof; diethylene glycol mono- or di-alkyl ethers; propylene glycol monoalkyl ethers and acetates thereof; dipropylene glycol mono- or di-alkyl ethers; methyl carbitol, butyl carbitol, butyl cellosolve acetate and carbitol acetate. 10

These organic solvents can be used individually or in combination of two or more thereof. The amount of this organic solvent used can be appropriately selected 15 according to the use end or the coating method.

In the case of applying the polymerizable composition of the present invention to uses where adhesion, hardness or the like is required, a commonly known inorganic filler may be used, if desired, for the purpose of improving various properties, and examples thereof include barium sulfate, barium titanate, powdered silicon oxide, finely powdered silicon oxide, amorphous silica, talc, clay, magnesium carbonate, calcium 20 carbonate, aluminum oxide, aluminum hydroxide and mica powder. Furthermore, if desired, commonly known additives may also be added, and examples thereof include a coloring agent such as Phthalocyanine Blue, Phthalocyanine Green, Iodine Green, Disazo Yellow, 25 Crystal Violet, titanium oxide, carbon black and naphthalene black; a silicone-type, fluorine-type or polymer-type defoaming agent and/or leveling agent; and an adhesion-imparting agent such as imidazole type, thiazole type, triazole type or silane coupling agent. 30

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Polymerization Method for Polymerizable Composition of the Present Invention:

5 The polymerizable composition of the present invention can be polymerized (cured) under irradiation of an active energy ray and/or under heat. The term "active energy ray" as used herein can be ultraviolet light, X-rays, an electron beam,  $\gamma$  rays or the like. In the case of using ultraviolet light, examples of the light source include a metal halide lamp, a mercury arc lamp, a xenon arc lamp, a fluorescent lamp, a carbon arc lamp, a tungsten-halogen duplicate lamp and sunlight.

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10 The irradiation conditions in the polymerization (curing) under irradiation of an active energy ray may be appropriately selected according to the light transmission which varies depending on the film thickness 15 of the polymerizable composition (the object to be cured) or the blending of a pigment or the like. In the case of curing a coating having a thickness of about 20  $\mu\text{m}$  using ultraviolet light, the dose may be from 1 to 2,000 20  $\text{mJ/cm}^2$ , preferably from 10 to 1,000  $\text{mJ/cm}^2$ . Furthermore, when the object to be cured contains a solvent, light can be irradiated after removing the solvent by heating or the like.

25 In the case of polymerization (curing) under heat, the conditions are generally a temperature of approximately from 50 to 400°C and a curing time of 5 seconds to 60 minutes, preferably approximately from 80°C to 250°C and from 10 seconds to 30 minutes.

30 After the irradiation of an active energy ray, if desired, the cured material may be heated. By this post-heating, the unreacted material remaining in the cured material may be reduced and the cure distortion generated in the cured material due to irradiation of an active 35 energy ray may be relaxed. As a result, hardness of the cured material may improve and in the case of a coating,

adhesion against the substrate to be coated may improve. This post-heating can be generally performed in an atmosphere temperature of 80 to 300°C for 5 seconds to 30 minutes.

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The polymerizable composition of the present invention can be applied to substrates such as metal, rubber, plastic, paper, wood, glass, ceramic and concrete.

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Examples of the use of the photosensitive resin composition of the present invention include a coating material, an adhesive, a sealing agent, a construction and building material, a laminate, electrical and electric parts, a photoresist, a solder resist, a layer insulating material for multilayer circuit boards, concrete structure repairing, a material for cast molding, a printing ink, a sealant, and a material for stereolithography.

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#### Examples

The present invention is described in greater detail below by referring to the Examples and the Comparative Examples, however, the present invention should not be construed as being limited to these Examples. Unless otherwise indicated, the term "parts" used in Examples and Comparative Examples is "parts by mass".

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Among the materials used in the Examples and the Comparative Examples, the commercially available products shown below were used in their original condition without passing through any purification.

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EPIKOTE 828: bisphenol A-type epoxy resin, produced by Yuka Shell Epoxy K.K.; XDO: 1,4-bis[(3-ethyl-3-oxetanylmethoxy)methyl]benzene, produced by Toa Gosei; EPIKOTE 152: phenol-novolak type epoxy resin, produced by

Yuka Shell Epoxy K.K.; KRM-2110: alicyclic base resin, produced by Asahi Denka Kogyo K.K.; SUN-AID SI-60L:  $\text{SbF}_6^-$  type sulfonium salt cationic polymerization initiator, produced by Sanshin Kagaku Kogyo; UVI-6990: triaryl sulfonium hexafluorophosphate salt-type cationic polymerization initiator, produced by UCC.

The compounds which are not commercially available were chemically synthesized by the present inventors.

That is, 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane and 6,7-epoxy-2-oxaspiro[3.5]nonane were synthesized by the present inventors with reference to U.S.Patent No.3,388,105. In particular, 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane was synthesized as below:

<Synthesis of 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane>

1) Synthesis of 6-methyl-3-cyclohexene-1,1-dimethanol

To a three-necked flask, 327 g of 2-methyl-4-cyclohexene-1-carboaldehyde, a Diels-Alder reaction product of butadiene and crotonaldehyde, 600 ml of methanol and 729 g of 37 % of formalin were charged, and then the temperature of this solution was raised to 60°C with stirring. Next, 252 g of KOH dissolved in 600 ml of distilled water was added dropwise thereto. After stirring for 7 hours, the reaction solution was concentrated under reduced pressure, which resulted in a residue with two phases. The oil phase concentrated to about 150 ml was washed with 300 ml of distilled water. After concentration under reduced pressure, 50 mg of 3,5-di(t-butyl)-4-hydroxytoluene (BHT) was added to the oil phase, and then vacuum distillation was carried out to yield 311 g (yield 82 %) of 6-methyl-3-cyclohexene-1,1-dimethanol as a colorless crystal.

35 2) Synthesis of 6-methyl-3-cyclohexene-1,1-dimethanol cyclic carbonate ester

To a three-necked flask, 310 g (1.99 mol) of 6-

5       methyl-3-cyclohexene-1,1-dimethanol, 894 g of dimethylcarbonate (DMC) and 0.93 g of potassium carbonate were charged, and was refluxed at a temperature raised to 90°C for 4 hours. The reaction solution was cooled to room temperature, and potassium carbonate was filtered off. After adding 120 mg of BHT, the remaining DMC and methanol were removed at a reduced pressure of 2 kPa (15mmHg), and then vacuum distillation was carried out to yield 326 g (yield 89.4 %) of 6-methyl-3-cyclohexene-1,1-dimethanol cyclic carbonate ester, which is in a form 10       of colorless crystals at room temperature.

3) Synthesis of 2-oxa-9-methylspiro[3.5]non-6-ene

15       To a three-necked flask, 321.15 g of 6-methyl-3-cyclohexene-1,1-dimethanol cyclic carbonate ester, 642 mg (0.2 mass %) of BHT, 1.93 g of LiCl were charged, and then stirred with heating at 275°C using a mantle heater. The reaction product was immediately drained out of the reaction system at a reduced pressure of about 8kPa (60mmHg), and heating was continued until no efflux 20       therefrom was observed. 600 mg of BHT was added to the product, and then vacuum distillation was carried out to yield 187 g (yield 71 %) of 2-oxa-9-methylspiro[3.5]non-6-ene as a transparent and colorless solution.

25       4) Synthesis of 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane  
30       50 g of 2-oxa-9-methylspiro[3.5]non-6-ene was dissolved in 150 ml of dichloromethane, and charged into a reaction vessel. 93.7 g of m-chloroperbenzoic acid suspended in 400 ml of dichloromethane was added dropwise over 1 hour, while keeping the temperature of the reaction solution under 40°C. The crystallized m-chlorobenzoic acid was filtered off, and washed thoroughly with cold dichloromethane. 15.0 g of potassium hydroxide was charged to the organic phase, and after 30 minutes, the crystallized crystal was filtered off, and washed with cold dichloromethane. The organic phase was 35

washed with 5 % of aqueous  $\text{NaHSO}_4$ , saturated saline, followed by concentration, to yield 38.1 g (yield 73.7 %) of 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane in a colorless semi-solid form.

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6,7-epoxy-2-oxaspiro[3.5]nonane was also synthesized by the present inventors with a similar procedure as above.

10 <Synthesis of 5-methyl-2-oxaspiro[3.5]nonane>

To a three-necked flask, 474 g of 2-methylcyclohexane-1,1-dimethanol, 405 g of dimethylcarbonate and 1.4 g of potassium carbonate were charged, heated in an oil bath at a temperature of 100°C with stirring, and the reaction was allowed to proceed for 14 hours while distilling out the formed methanol from the reaction system under atmospheric pressure. Finally, the pressure within the reaction vessel was reduced to 10 mmHg, yielding a carbonate ester in a yield of 95%.

25 The resulting cyclic carbonate ester was heated to 250°C, and the reaction was allowed to proceed for 10 hours while distilling out the formed carbonic acid gas from the reaction system via the top of a cooling instrument. This reaction solution was subjected to a distillation purification to yield 230 g of 5-methyl-2-oxaspiro[3.5]nonane.

30 <Synthesis of spiro[5,6-epoxynorbornane-2,3'-oxetane]>

This compound was synthesized by the present inventors in exactly the same manner as the synthesis of 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane, except for using a Diels-Alder reaction product of cyclopentadiene and acrolein in place of the Diels-Alder reaction product of butadiene and crotonaldehyde.

Method for Evaluating Polymerizability and Physical Properties of Cured Material:

In the evaluation of polymerizability and physical properties of cured material, the reaction conversion, 5 the surface hardness and the pencil scratch value of cured material were determined and used as indices. Specifically, the values were determined by the following operations.

10 1) Reaction Conversion

The reaction conversion was calculated from the variation of infrared absorption spectrum before and after the polymerization of the polymerizable composition coated on a silicon wafer to a thickness of 10  $\mu\text{m}$ .

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The infrared absorption spectrum was measured by the transmission method using an FTIR measuring apparatus (Fourier transformation infrared spectrophotometer, Model VALOR-III, manufactured by Nihon Bunko K.K.).

20

The reaction conversion was determined from the amount of the variation, before and after the irradiation, in the absorbance of the characteristic absorption peak in the infrared absorption spectrum, more 25 specifically, for oxetanyl group, the characteristic absorption peak at  $980\text{ cm}^{-1}$ ; for alicyclic epoxy group, the characteristic absorption peak at from  $789$  to  $798\text{ cm}^{-1}$ ; and for glycidyl group, the characteristic absorption peak at  $773\text{ cm}^{-1}$ , respectively, according to the 30 following formula:

Conversion =

(1 - absorbance of characteristic absorption peak after irradiation/absorbance of characteristic absorption peak before 35 irradiation)  $\times 100$  (%)

In Examples 6 to 8, the average of reaction conversion values of the oxetanyl group and the epoxy group present within the same molecule was used as the conversion. In Comparative Examples 3 to 8, the 5 conversion was calculated from the peak of characteristic absorption (glycidyl group in EPIKOTE 828, oxetanyl group in XDO, and alicyclic epoxy group in KRM-2110) of the material used for comparison.

10 2) Surface Hardness

The surface hardness of the cured materials were measured using a hardness tester manufactured by Colman Co. (GYZJ934-1). The measurement temperature was 23°C.

15 3) Pencil Scratch Value of Cured Material

The cured coatings of the polymerizable compositions on silicon wafers were measured in accordance with JIS K5400.

20 Thermopolymerization of Composition:  
(Example 1)

To 100 parts of 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane, 5 parts of an acid generation-type cationic polymerization initiator "SUN-AID SI-60L, 25 produced by Sanshin Kagaku Kogyo K.K." was added and thoroughly stirred to obtain a thermopolymerizable composition. Laminates, each comprising a 2 mm-thick glass plate, wherein a 2 mm-thick rubber spacer having a 5 cm diameter bored hole is placed thereon and a 30 thermopolymerizable composition being poured into the bored hole, and another glass plate laminated to the former plate in order to prevent the entry of air, were provided. Two units of this device were prepared and left standing in an oven at 120°C. One was taken out 35 from the oven after 10 minutes and another was taken out after 30 minutes to obtain cured materials having a diameter of 5 cm and a thickness of 2 mm. After the

5       cured materials had cooled to room temperature, the temperature was adjusted to 23°C and the surface hardness of each cured material with a curing time of 10 minutes or 30 minutes was measured. The results obtained are shown in Table 1.

(Example 2)

10       The operation was performed in exactly the same manner as in Example 1 except for using 6,7-epoxy-2-oxaspiro[3.5]nonane in place of 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane, and then the surface hardness of each cured material was measured. The results obtained are shown in Table 1.

15       (Example 3)

20       The operation was performed in exactly the same manner as in Example 1 except for using spiro[5,6-epoxynorbornane-2,3'-oxetane] in place of 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane, and then the surface hardness of each cured material was measured. The results obtained are shown in Table 1.

(Example 4)

25       The operation was performed in exactly the same manner as in Example 1 except for using a mixture of 90 parts of 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane and 10 parts of EPIKOTE 828 (produced by Yuka Shell Epoxy K.K.) in place of 100 parts of 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane, and then the surface hardness of each cured material was measured. The results obtained are shown in Table 1.

(Example 5)

35       The operation was performed in exactly the same manner as in Example 1 except for using a mixture of 80 parts of 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane, 10 parts of EPIKOTE 828 (produced by Yuka Shell Epoxy K.K.)

and 10 parts of 5-methyl-2-oxaspiro[3.5]nonane in place of 100 parts of 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane, and then the surface hardness of each cured material was measured. The results obtained are shown in Table 1.

5

(Example 6)

The operation was performed in exactly the same manner as in Example 1 except for using a mixture of 10 parts of 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane and 90 parts of EPIKOTE 152 (produced by Yuka Shell Epoxy K.K.) in place of 100 parts of 7,8-epoxy-2-oxa-5-methyl spiro[3.5]nonane, and then the surface hardness of each cured material was measured. The results obtained are shown in Table 1.

15

(Comparative Examples 1 and 2)

The operations were performed in exactly the same manner as in Example 1 except for using a heat polymerizable composition having the formulations shown in Table 1, and then the surface hardness of each cured material was measured. The results obtained are shown in Table 1.

As is apparent from Table 1, the polymerizable compositions of the present invention in Examples 1 to 5 achieved the final hardness with heating for about 10 minutes, and this reveals that each composition had rapidly polymerized and cured.

Also, it is seen from the comparison between Example 6 and Comparative Example 2 that the curing rate of epoxy resin can be improved by blending about 10 parts by mass of the alicyclic alkane (a) having at least one oxetanyl group and at least one epoxy group within the molecule of the present invention, into the epoxy resin.

Table 1

Blending of Polymerizable Composition and Surface  
Hardness of Cured Material

Name of Compound	Examples						Comparative Examples	
	1	2	3	4	5	6	1	2
7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane	100	0	0	90	80	10	0	0
6,7-epoxy-2-oxaspiro[3.5]nonane	0	100	0	0	0	0	0	0
Spiro[5,6-epoxynorbornane-2,3'-oxetane]	0	0	100	0	0	0	0	0
5-Methyl-2-oxaspiro[3.5]nonane	0	0	0	0	10	0	0	0
EPIKOTE 828	0	0	0	10	10	0	0	0
XDO	0	0	0	0	0	0	100	0
EPIKOTE 152	0	0	0	0	0	90	0	100
SUN-AID SI-60L	5	5	5	5	5	5	5	5
Surface hardness (hard) after heat-curing for 10 minutes	20	18	22	20	18	20	0	5
after heat-curing for 30 minutes	25	24	27	25	24	25	2	10

Photopolymerization of Composition

5 (Example 7)

To 100 parts of 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane, 5 parts of photoacid generation-type cationic polymerization initiator "UVI-6990" produced by Union Carbide K.K. was added and thoroughly 10 stirred to obtain a photopolymerizable composition.

Subsequently, this composition was coated on a silicon wafer to a thickness of 10  $\mu\text{m}$  and the conversion of the reactive groups was determined following the 15 irradiation with a high-pressure mercury lamp at 1.45  $\text{mW/cm}^2$  for 10 seconds and for 30 seconds.

Furthermore, in accordance with the measurement method of 3) Pencil Scratch Value above, the pencil 20 scratch value of the cured coating on the silicon wafer obtained by the irradiation for 30 seconds was measured.

The results obtained are shown together in Table 2.

(Example 8)

5 The operation was performed in exactly the same manner as in Example 7 except for using a mixture of 90 parts of 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane and 10 parts of EPIKOTE 828 (produced by Yuka Shell Epoxy K.K.) in place of 100 parts of 7,8-epoxy-2-oxa-5-  
10 methylspiro[3.5]nonane, and then the conversion and the pencil scratch value were measured. The results obtained are shown in Table 2.

(Example 9)

15 The operation was performed in exactly the same manner as in Example 7 except for using a mixture of 50 parts of 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane and 50 parts of EPIKOTE 828 (produced by Yuka Shell Epoxy K.K.) in place of 100 parts of 7,8-epoxy-2-oxa-5-  
20 methylspiro[3.5]nonane, and then the conversion and the pencil scratch value were measured. The results obtained are shown in Table 2.

25 From the results of Examples 7 to 9, it was demonstrated that the conversions of oxetanyl group and alicyclic epoxy group were more improved, and a more rapid curing rate was achieved by blending an epoxy compound as a constituent element of the composition of the present invention, to the alicyclic alkane having at least one oxetanyl group and at least one epoxy group  
30 within the same molecule.

(Comparative Examples 3 to 5)

35 The operation was performed in exactly the same manner as in Example 7 except for using a photo-polymerizable composition having formulations shown in Table 2, and the conversions and the pencil scratch values were measured. The results obtained are shown in

Table 2.

As is apparent from Table 2, following the light irradiation, the reactive groups of the polymerizable compositions of the present invention in Examples 7 to 9 decreased rapidly, namely, they polymerized rapidly as compared with the Comparative Examples 3 to 5, revealing that the compositions can be polymerized and cured within a short period of time.

10

## [Table 2]

## Blending and Conversion of Polymerizable Composition and Pencil Scratch Value of Cured Material

Name of Compound	Examples			Comparative Examples		
	7	8	9	3	4	5
7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane	100	90	50	0	0	0
EPIKOTE 828	0	10	50	0	0	0
XDO	0	0	0	100	0	0
EPIKOTE 152	0	0	0	0	100	0
KRM-2110	0	0	0	0	0	100
UVI-6990	5	5	5	5	5	5
Conversion (%) after irradiation of 10 seconds	63	70	87	20	24	13
after irradiation of 30 seconds	82	83	91	46	40	29
Pencil scratch value	3H	3H	3H	3H	3H	3H

15

In the Examples and Comparative Examples of each Table, compounds shown below were used.

EPIKOTE 828: bisphenol A-type epoxy resin, produced by Yuka Shell Epoxy K.K.

XDO: 1,4-bis[(3-ethyl-3-oxetanylmethoxy)methyl]benzene, produced by Toa Gosei

EPIKOTE 152: phenol-novolak type epoxy resin, produced by Yuka Shell Epoxy K.K.

KRM-2110: alicyclic base resin, produced by Asahi Denka Kogyo K.K.

20

SUN-AID SI-60L:SbF<sub>6</sub><sup>-</sup> type sulfonium salt cationic

25

polymerization initiator, produced by  
Sanshin Kagaku Kogyo K.K.

UVI-6990: triaryl sulfonium hexafluorophosphate  
salt-type cationic polymerization  
5 initiator, produced by UCC.

### Industrial Applicability

The present invention provides a cationic  
polymerizable composition exhibiting high activity (rapid  
10 curability) under heat and/or under irradiation of an  
active energy ray and also provides a cured material  
thereof and a method for producing the cured material.  
The present invention can be applied to the coating on  
paper or plastic having poor heat resistance and further  
15 can be applied to a coating, an adhesive, a sealing  
agent, a construction and building material, a laminate,  
electrical and electronic parts, a photoresist, a solder  
resist, a layer insulating film for multilayer circuit  
boards, concrete structure repairing, or a material for  
20 cast molding, a printing ink, a sealant, and a material  
for stereolithography.

It will be appreciated by those skilled in the art  
that while the invention has been described above in  
25 connection with particular embodiments and examples, the  
invention is not necessarily so limited and that numerous  
other embodiments, examples, uses, modifications and  
departures from the embodiments, examples and use may be  
made without departing from the inventive scope of this  
30 application.

CLAIMS

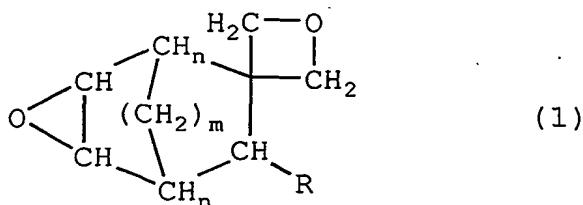
1. A polymerizable composition comprising (a) an alicyclic alkane having at least one oxetanyl group and at least one epoxy group within the same molecule and (b)

5 a compound capable of initiating cationic polymerization under irradiation of an active energy ray and/or under heat.

2. The polymerizable composition according to claim 1, wherein said compound (b) is a compound capable 10 of generating an acid under irradiation of an active energy ray and/or under heat, and thereby initiating the cationic polymerization.

3. The polymerizable composition according to claim 2, wherein said compound (b) is one or more 15 compound selected from the group consisting of sulfonium salts, iodonium salts and diazonium salts.

4. The polymerizable composition according to claim 1, wherein said alicyclic alkane (a) is a compound represented by formula (1):



wherein R represents a hydrogen atom or a methyl group, m represents an integer of 0 to 2, and n is 2 when m is 0 and otherwise n is 1.

25 5. The polymerizable composition according to claim 4, wherein said alicyclic alkane (a) is 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane.

30 6. The polymerizable composition according to claim 1, which comprises a compound (c) that can be cationic polymerized by said compound (b), wherein said compound (c) is other than said compound (a).

7. The polymerizable composition according to claim 6, wherein at least one of said compounds (c) is a

compound (c-1) having one or more epoxy group that can be cationic polymerized by said compound (b).

8. The polymerizable composition according to claim 6, wherein at least one of said compounds (c) is a compound (c-2) having one or more oxetanyl group that can be cationic polymerized by said compound (b).

9. The polymerizable composition according to claim 1, wherein said alicyclic alkane (a) is blended in an amount of 5 to 100 mass % based on the total mass of the polymerizable composition minus the mass of said compound (b).

10. The polymerizable composition according to claim 7, wherein said compound (c-1) is blended in an amount of 5 to 95 mass % based on the total mass of the polymerizable composition minus the mass of said compound (b).

11. The polymerizable composition according to claim 8, wherein said compound (c-2) is blended in an amount of 5 to 95 mass % based on the total mass of the polymerizable composition minus the mass of said compound (b).

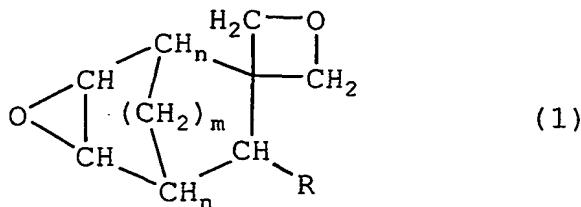
12. A cured material obtained by polymerizing a polymerizable composition comprising (a) an alicyclic alkane having at least one oxetanyl group and at least one epoxy group within the molecule and (b) a compound capable of initiating cationic polymerization under irradiation of an active energy ray and/or under heat.

13. The cured material according to claim 12, wherein said compound (b) is a compound capable of generating an acid under irradiation of an active energy ray and/or under heat, and thereby initiating the cationic polymerization.

14. The cured material according to claim 13, wherein said compound (b) is one or more compound selected from the group consisting of sulfonium salts, iodonium salts and diazonium salts.

15. The cured material according to claim 12,

wherein said alicyclic alkane (a) is a compound represented by formula (1):



5        wherein R represents a hydrogen atom or a methyl group, m represents an integer of 0 to 2, and n is 2 when m is 0 and otherwise n is 1.

10       16. The cured material according to claim 15, wherein said alicyclic alkane (a) is 7,8-epoxy-2-oxa-5-methylspiro[3.5]nonane.

15       17. The cured material according to claim 16, which comprises a compound (c) that can be cationic polymerized by said compound (b), wherein said compound (c) is other than said compound (a).

20       18. The cured material according to claim 17, wherein at least one of said compound (c) is a compound (c-1) having one or more epoxy group that can be cationic polymerized by said compound (b).

25       19. The cured material according to claim 17, wherein at least one of said compound (c) is a compound (c-2) having one or more oxetanyl group that can be cationic polymerized by said compound (b).

20       20. The cured material according to claim 12, wherein said alicyclic alkane (a) is blended in an amount of 5 to 100 mass% based on the total mass of the polymerizable composition minus the mass of said compound (b).

30       21. The cured material according to claim 18, wherein said compound (c-1) is blended in an amount of 5 to 95 mass % based on the total mass of the polymerizable composition minus the mass of said compound (b).

22. The cured material according to claim 19, wherein said compound (c-2) is blended in an amount of 5

to 95 mass % based on the total mass of the polymerizable composition minus the mass of said compound (b).

23. A method for manufacturing a cured material from a polymerizable composition comprising (a) an alicyclic alkane having at least one oxetanyl group and at least one epoxy group within the same molecule and (b) a compound capable of initiating cationic polymerization under irradiation of an active energy ray and/or under heat, said method comprising initiating the polymerization under irradiation of an active energy ray and/or under heat.

10 24. A method for manufacturing the cured material according to any of claims 12 to 22, comprising initiating the polymerization under irradiation of an active energy ray.

15 25. The method according to claim 24, wherein said active ray is ultraviolet light.

20 26. A method for manufacturing the cured material according to any of claims 12 to 22, comprising initiating the polymerization under heat.

27. A method for manufacturing the cured material according to any of claims 12 to 22, comprising initiating the polymerization under irradiation of an active energy ray and then heating the composition.

25 28. The method according to claim 27, wherein said active ray is ultraviolet light.

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(71) Applicant (for all designated States except US): SHOWA DENKO K. K. [JP/JP]; 13-9, Shiba Daimon 1-chome, Minato-ku, Tokyo 105-8518 (JP).

(72) Inventors; and

(75) Inventors/Applicants (for US only): WATANABE, Takeo [JP/JP]; C/O CENTRAL RESEARCH LABORATORY, SHOWA DENKO K.K., 1-1, Ohnodai 1-chome, Midori-ku, Chiba-shi, Chiba 267-0056 (JP). SATO, Takashi [JP/JP]; C/O CENTRAL RESEARCH LABORATORY, SHOWA DENKO K.K., 1-1, Ohnodai 1-chome, Midori-ku, Chiba-shi, Chiba 267-0056 (JP). TAGOSHI, Hirotaka [JP/JP]; C/O CENTRAL RESEARCH LABORATORY, SHOWA DENKO K.K., 1-1, Ohnodai 1-chome, Midori-ku, Chiba-shi, Chiba 267-0056 (JP).

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(54) Title: POLYMERIZABLE COMPOSITION, CURED MATERIAL THEREOF AND METHOD FOR MANUFACTURING THE SAME

(57) Abstract: The present invention provides a polymerizable composition comprising (a) an alicyclic alkane having at least one oxetanyl group and at least one epoxy group within the same molecule and (b) a compound capable of initiating cationic polymerization under irradiation of an active energy ray and/or under heat exhibits high activity in the photo- or heat-cationic ring-opening polymerization exhibiting high activity, e.g., rapid polymerizability, rapid curability, under irradiation of an active energy ray and/or under heat. The present invention also provides a polymerization product thereof.

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## INTERNATIONAL SEARCH REPORT

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A. CLASSIFICATION OF SUBJECT MATTER  
 IPC 7 C08G65/18 C08G59/68

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Minimum documentation searched (classification system followed by classification symbols)  
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Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, PAJ, WPI Data, CHEM ABS Data

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 3 457 193 A (TINSLEY SAMUEL W ET AL) 22 July 1969 (1969-07-22)  example 18 ---	1,2,4,5, 12,13, 15,16, 23,26
A	GB 2 310 211 A (KANSAI PAINT CO LTD) 20 August 1997 (1997-08-20) claims 1,5; table 1 ---	1-28
A	PATENT ABSTRACTS OF JAPAN vol. 1999, no. 08, 30 June 1999 (1999-06-30) & JP 11 061034 A (KANSAI PAINT CO LTD), 5 March 1999 (1999-03-05) abstract ---	1-28

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## Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2  
 NL - 2280 HV Rijswijk  
 Tel. (+31-70) 340-2040, Tx. 31 651 epo nl.  
 Fax: (+31-70) 340-3016

## Authorized officer

O'Sullivan, T

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## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 897 710 A (ESPE DENTAL AG) 24 February 1999 (1999-02-24) claim 1 page 4 ---	1-28
A	US 3 388 105 A (JOACHIM GALINKE ET AL) 11 June 1968 (1968-06-11) cited in the application claim 1 ---	1-28
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Information on patent family members

International Application No

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